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Synergistic effects of metal nanoparticles and a phenolic uncoupler using microdroplet-based two-dimensional approach

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A droplet-based microfluidic technique for testing multiple reagent concentrations is presented. We used this experimental approach to study combined effects of gold (AuNP) and silver nanoparticles (AgNP) with the phenolic uncoupler 2,4-dinitrophenol (DNP) with respect to the growth of *Escherichia coli*. In order to evaluate the toxicity of binary mixtures, we first encapsulated the *E. coli* cells and particle mixtures inside the microdroplets using PEEK (polyetherketone) modules. Two-dimensional concentration spaces with about 500 well separated droplets were addressed. We subsequently analyzed the cell growth, the viability and the autofluorescence intensity (metabolic activity) of the bacteria with a micro-flow-through fluorometer and photometer. Dose-dependent synergistic effects were found for the binary mixture of AgNPs and DNP, which indicated a stronger interaction in the mixture than it was expected from effect summation. For the binary mixture of DNP and AuNPs in non-toxic concentrations, we found only weak synergistic effects at low DNP concentrations. Furthermore, the non-toxic tested AuNPs causes effect summation in the binary mixture with the phenolic uncoupler. In general, we demonstrated the efficiency of a droplet-based microfluidic system for fast high-throughput screenings of binary and multiple mixtures. This work also confirmed the relevance of highly resolved droplet-based assays for the miniaturization of ecotoxicological aquatic test systems.

Introduction

Metal nanoparticles have attracted increasing scientific as well as technical interest because of their unique set of physical and chemical properties. Due to their spectral features as well as their electronic and catalytic behaviour, they are part of a material class with several promising applications.^{1–4} A key feature of nanoparticles is that their properties can be tuned by the judicious choice of the core material and are strongly influenced by their size, shape and surface state.⁵

The toxicological implications of metal nanoparticles are comparatively simple to understand if the particles are completely corroded and the resulting metal ions are released into the solution. The effect of the corrosion products is mainly given by the concentration of free ions, the solubility of salts formed by the ions and their coordination chemistry.⁶ The situ-

ation becomes much more complicated if the nanoparticles are more or less stable. In this case, the interaction of the material with tissues, cells and membranes is additionally controlled by shape and size as well as their mobility, interaction potential, aggregation behaviour, their electrical charges and chemical surface state. In particular, the last two mentioned parameters are strongly affected by the chemical environment. Since nanoparticles possess a very high specific surface area, any chemical species of a liquid matrix can interact with this surface by unspecific adsorption, specific binding and other chemical reactions. Therefore, it is reasonable to expect that the effects of nanoparticles on cells are strongly dependent on the chemical composition of the matrix.^{7,8}

The high chemical activity of nanoparticles can be explained by the presence of a large number of surface binding sites with high local energy. In particular, small spherical particles show unexpected chemical properties due to active sites on crystal edges and corners resulting from the presence of small-sized crystal planes of high indices. The constitutional properties of a certain type of nanoparticle lead to a specific chemical behavior. As a result, less reactive or even inert materials as some noble metals are converted into highly reactive materials with

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Environmental impact

There are a huge number of consumer products which contain nanomaterials, nowadays. The environmental impact of the released nanoparticles is not yet predictable. Even less is known about the effect of nanoparticles in chemical mixtures, which describe the real status in our environment. In this work we report the application of a droplet-based microfluidic set-up opens an efficient way for evaluating mixture effects of pollutants and interferences of nanoparticles with other substances. The study shows the effects of silver and gold nanoparticles in binary mixture with a phenolic uncoupler on the bacteria *Escherichia coli*. We believe that the potential of droplet-based microfluidic allows future high resolved ecotoxicological screenings of mixtures toxicity within the nanolitre range.

surprising effects if these materials are nanosized. This is not only the case for platinum, palladium and silver with their well-known catalytic activities, but also for gold.^{9,10} For example, it was found that gold nanoparticles can catalyze organic reactions as partial oxidation of olefins with high specificity.¹¹

The evaluation of the toxic potential of nanoparticles must reconsider the specificity of nanoparticles and their chemical state. Therefore, the determination of MIC (minimum inhibitory concentration) or EC50 (effective concentration 50%) values must be strongly related to the matrix conditions.¹² Besides, contaminated systems will probably be polluted with mixtures of pollutants, leading to possible interactions and changes in their effectiveness. This results in a combinatorial problem and demands combinatorial screenings for the toxicological properties of nanoparticles. In this case, the well-introduced classical toxicological information from a minimum of cells and chemicals are required.

The microdroplet-based flow offers some important advantages over techniques based on micro- and nanotiterplates or conventional ecotoxicological methods for solving the problem of screening a complex parameter field. Besides the high throughput of small volumes, the significantly reduced evaporation, the reduced wall contact and the mixing by segment-internal convection make the microdroplet-based flow ideal for screening operation. Thereby, an aqueous cultivation phase is subdivided into a lot of small and well separated aliquots by injection into a water-immiscible inert liquid.^{13,14}

Microdroplets are generated readily by microfluidic actuation. In this manner, microdroplet sequences with hundreds or thousands of well-defined aliquots can be generated in only a few seconds or minutes. The volumes and the concentrations of the separated liquids can be set with astonishing precision up to picolitre and picomolar range.¹⁵

It was shown that this technique can be applied for analytical purposes, chemical synthesis, cell cultivation and screenings with microorganisms as well as multicellular systems.^{8,16–20} Information about cell density or physiological activities inside the fluid droplets can be extracted non-invasively by microphotometric or microfluorimetric methods. Thus, this method is well suited for quantitative measurements with small sample volumes and for highly parallelized screenings. In this paper, the droplet-based flow principle is introduced for studying synergistic effects of colloidal suspensions of gold and silver nanoparticles with 2,4-DNP which is known to be an important pollutant.

Materials and methods

Chemicals

The following chemicals were used: 2,4-dinitrophenol (DNP, Acros), perfluoromethyldecalin (PP9, F2 Chemicals Ltd), cochineal red A (VWR), indigo carmine (Fluka Analytical), silver nitrate (AgNO₃, Merck-Schuchardt), 25% ammonia solution (Merck-Schuchardt), D-(+)-glucose (dextrose, Carl Roth), nitric acid (HNO₃, Merck-Schuchardt), sodium hydroxide (NaOH, Merck-Schuchardt), hydrogen tetrachloroaurate trihydrate (HAuCl₄, Carl Roth), L-(+)-ascorbic acid (C₆H₈O₆, Merck) and ammonium chloride (NH₄Cl, Merck). The synthetic medium was composed of the following components: 0.56 g K₂HPO₄, 0.3

g KH₂PO₄, 0.012 g MgSO₄ × 7H₂O, 0.1 g NH₄Cl in 90 mL distilled water and 10 mL 0.1 g L⁻¹ glucose solution.

Biological material and preparation of samples

The experiments were performed with the Gram-negative bacteria *Escherichia coli* (strain RV308 provided by the Hans-Knöll institute (HKI), Jena, Germany). The bacterial pre-culturing was carried out in synthetic medium under stirring for 24 h at 37 °C.

The inoculum cell density was adjusted to 100 000 cells mL⁻¹ by the adequate dilution in synthetic medium. The cell cultures had been attached in the lag-phase and the early exponential phase of bacterial growth. The indicator dyes cochineal red A (0.004%) and indigo carmine (0.01%) were added to the cell suspension. The initial concentration of the effector suspensions for AuNP, AgNP and 2,4-DNP was adjusted to 20 pmol (particle) L⁻¹, 0.3 pmol (particle) L⁻¹ and 300 μmol L⁻¹.

For the detection of the concentration distribution inside the droplets cochineal red A and indigo carmine had been added as indicator dyes to the effectors. The distribution of the effectors in the droplets is expected to be same as the indicator dyes. To distinguish between the concentration setting of the different effectors within the droplets two different indicator dyes had been used.

For the monitoring of the concentration setting inside the nanolitre droplets, the indicator dye cochineal red A (0.004%) was added to the AuNP and AgNP suspension as well as the indicator dye indigo carmine (0.01%) to the DNP-solution.

Preparation of silver and gold nanoparticles

The synthesis of silver nanoparticles was performed according to the specification listed in Kvitek *et al.*²¹ using D-(+)-glucose for the reduction of the complex anion [Ag(NH₃)₂]⁺ in alkali media [pH 12.5] at 23 °C. The synthesis of gold nanoparticles was carried out by the reduction of tetrachloroaurate with ascorbic acid at 23 °C. In this case, 2 mL of 1 mmol L⁻¹ of HAuCl₄ and 0.1 mL of 0.1 mol L⁻¹ ascorbic acid had been mixed and diluted with 5 mL distilled water.

The number and weight related size distribution of the synthesized metal nanoparticles was characterized by differential centrifugal sedimentation (DCS, using DC 20000, CPS Instruments Inc., USA (see Table 1)). The dimensions and shapes of the gold and silver nanoparticles were determined by scanning electron microscopy (SEM, JSM 6700, JEOL USA) and transmission electron microscopy (TEM, Philips Tecnai 20 S-TWIN)

Table 1 Average particle size and initial concentration of the prepared metal nanoparticles

Metal nanoparticles	Particle size ^a / half-width/nm	Particle size ^a / half-width/nm	Initial concentration of metal nanoparticles (related to particle type A)/pmol L ⁻¹
AuNP	30/11.5	3/0.3	10
AgNP	21.2/9	2.8/1.9	0.15

^a The two different size distributions found for both types of nanoparticles determined by DCS.

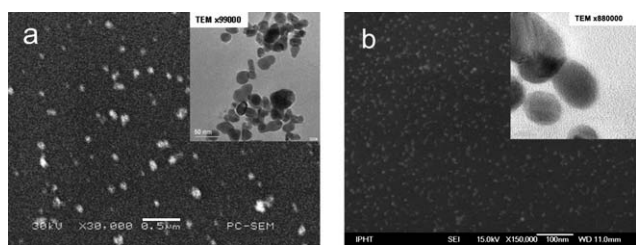


Fig. 1 SEM and TEM images of silver (a) and gold nanoparticles (b).

(Fig. 1). The influence of the synthetic cultivation medium on gold as well as the silver nanoparticles was investigated by UV/VIS spectroscopy (Specord 2000, Analytik Jena AG, Germany). The composition of the reagents in the UV-cuvette had been set identical to the microdroplets composition. After mixing the colloidal suspension and the cultivation buffer without cells the optical spectra of the colloidal suspensions were registered immediately within 24 h and 48 h after incubation at 37 °C. Control experiments with pure water were executed in parallel.

The influence of nanoparticles on the light scattering was corrected by setting the measurement signals against the reference signal ($t = 0$ h). The ion strength during the nanoparticle synthesis was given by the concentration of dissociation products of the applied reactants. The size of nanoparticles is effected by the ratio of nucleation and particle growth. It can be controlled by the reactant concentration.

Instrumentation

A microfluidic system was used for the monitoring of single and two-dimensional dose–response relationships (Fig. 2). The generation of the microdroplet sequences was realized in a 7-Port-Peek Manifold (Upchurch, Washington) by targeted dosing of bacteria suspension and chemical substances into a carrier liquid (perfluoromethyldecalin, PP9). Therefore, the manifold was connected through Teflon® tubes (0.5 mm ID, Bohlender

GmbH, Grünsfeld, Germany) with a computer-controlled syringe pump equipped with six dosing units (Cetoni GmbH, Korbueßen, Germany), and syringes (ILS, Stützerbach, Germany) with volumes of 500 µL and 2500 µL for the samples and the carrier liquid, respectively. For homogeneous loading of the cells inside the microdroplets, an integrated syringe stirring system (neMIX, Cetoni GmbH) was used at 650 rpm (Fig. 2a).

The fluid flow was controlled using a software based on LabVIEW™ (National Instruments) for the single and two-dimensional concentration adjustment inside the microdroplets. The binary mixtures were performed in a $(n^2 + n)/2$ design inside the 7-Port Manifold by a flow rate controlled variation of substance A, substance B and cultivation medium whereas the flow rate of the cell suspension was kept constant. For each concentration, six to seven microdroplets were generated, whereas for the positive control (only cells) the first 10 microdroplets were addressed. Each sequence comprised 400 to 500 microdroplets with a volume of about 600 nL. The flow rates of the carrier liquid and samples were regulated at 40 µL min⁻¹ and 4 µL min⁻¹ and the total flow rate for the generation process was set to 50 µL min⁻¹.

The experimental data were set according to the concentration addition model irrespective of the underlying modes of action and illustrate in isobolograms. The EC(50) values from the dose–response curves of the single substances were obtained by appropriate fits of the data points. The term “synergism” was used to describe stronger effects of the mixtures as expected from concentration addition.

The monitoring of the microdroplet signal was performed directly inside a transparent FEP-tubing (Upchurch). The combined application of a micro-flow-through photometer and fluorometer allowed the simultaneous measurement of the scattered light and fluorescence intensity (Fig. 2b). Two light emitting diodes (LED, Agilent, Santa Clara CA, United States) with a peak wavelength of 505 nm and 610 nm were used for the scattering measurement and absorbance detection of the indicator dyes.

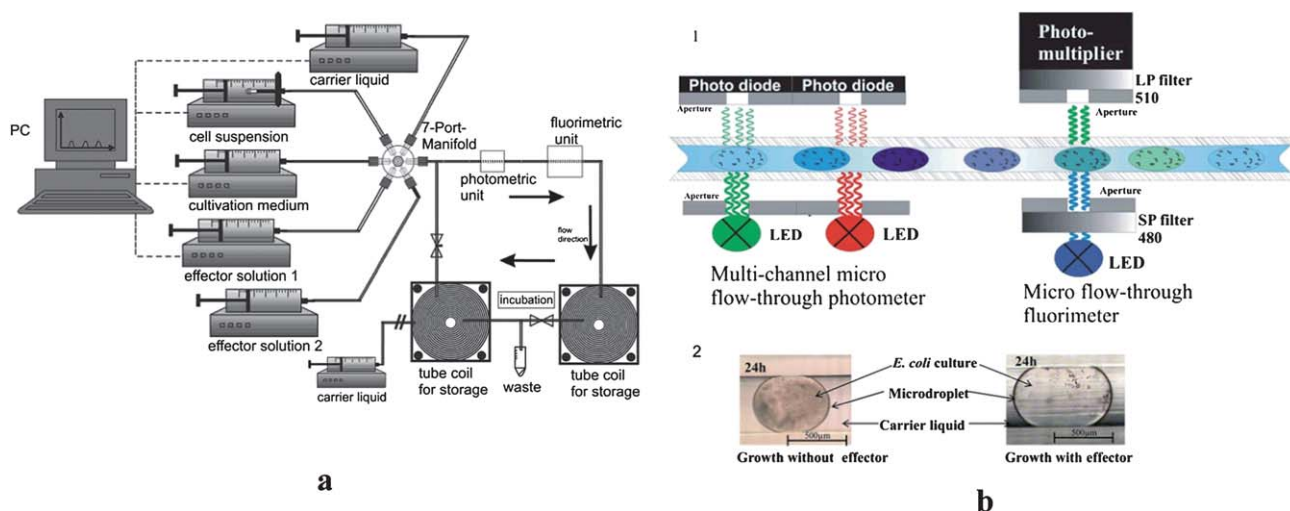


Fig. 2 Experimental set-up for the generation and the analysis of microdroplets for one- and two-dimensional toxicological screenings. (a) Scheme of the experimental set-up. (b) 1. Pathway for detection of microdroplet sequences. 2. Growth of *E. coli* in microdroplets with and without the presence of a binary mixture.

For detecting the autofluorescence signal of *E. coli* the excitation (LED, 470 nm, CML, Bayern, Germany)/emission wavelength of the micro-flow-through fluorometer were set to 480 nm/510 nm through appropriate filters (Laser Components GmbH, Olching, Germany). An aperture with a split width of 1 mm focused the light beam onto the tube.

A tube coil was used to store the generated segment sequences between every measurement.

Results and discussion

1. Interaction between cultivation buffer (synthetic medium) and nanoparticles in the absence of cells

The stability of the colloidal systems depends on the electrical surface state of the particles, which is influenced by ligands, pH and ionic strength of the suspension.²² Therefore, the effect of the cultivation medium on the nanoparticle suspensions was checked in the absence of cells. Changes in the colloidal suspension are indicated by a response of the optical absorption to the changes of the surrounding medium.

All samples showed a change of the optical spectra over time. The silver plasmon absorption at 410 nm shows a strong decrease in absorbance and a hypsochromic shift after one and two days incubation if the colloidal suspension was mixed in the cultivation medium. And a new, small absorption band appeared at about 350 nm. On the other hand, the plasmon absorption band of gold nanoparticles is immediately affected by the cultivation medium. The shape of optical spectra changed instantaneously after mixing. The behavior of the nanoparticles can be interpreted by an aggregation and sedimentation of nanoparticles and adsorption at the surface of the cuvette wall. In the case of silver the strong reduction of original absorption and the arising short-wave length band could also be due to an oxidation of silver nanoparticles.

In conclusion, it can be expected that the effective concentrations of metal nanoparticles and the size of the nanoparticles are lower/bigger in the microtoxicological screenings as estimated from the suspension ratio in the preparation. So, the given nanoparticle concentrations are upper limits for the presence of metal nanoparticles.

2. Microtoxicological screening of the binary mixture of gold nanoparticles and DNP inside microdroplets

Gold nanoparticles are regarded to be non-toxic in the past because bulky gold was found to be bio-inert.²³ However, some pharmaceutical studies indicate a physiological activity of gold nanoparticles.²⁴ This motivates detailed investigations on the toxicity of noble metal nanoparticles such as gold and in particular those in the presence of other toxic substances.

The toxic effect of gold nanoparticles was checked in the presence of 2,4-dinitrophenol (DNP) as an example of a well-known cell-damaging aromatic compound. In this case, the complete two-dimensional concentration space between 0 and 10 pM gold nanoparticles and 0 and 150 $\mu\text{mol L}^{-1}$ DNP was covered by a droplet-sequence containing approximately 500 single and well-separated microdroplets. Additionally, one-dimensional screenings of AuNPs and DNP were carried out in microdroplets (Fig. 3). The growth of bacteria was monitored by

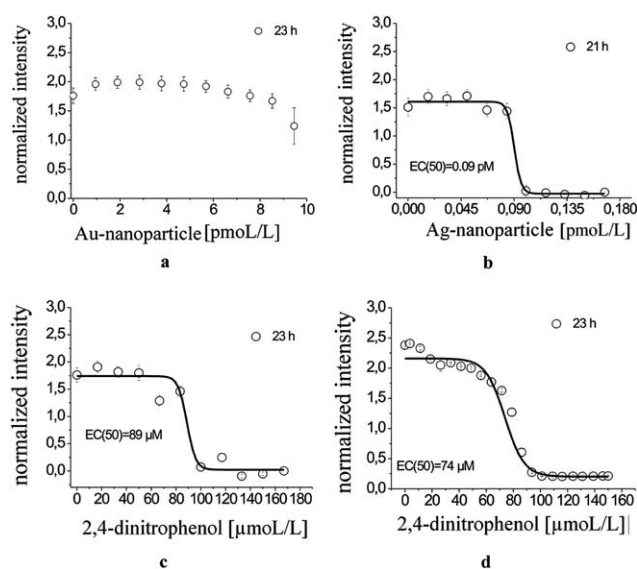


Fig. 3 Dose-response curves for the single effectors DNP and AuNPs and their binary mixture on the autofluorescence intensity of *E. coli* after 24 h cultivation in microdroplets. The normalized intensity† was determined from nearly 500 microdroplets for two-dimensional and 200 for one-dimensional screening respectively. (a) Single effect of AuNPs determined by two-dimensional screening at a DNP concentration of 0 $\mu\text{mol L}^{-1}$. (b) Single effect of AgNPs determined by two-dimensional screening at a DNP concentration of 0 $\mu\text{mol L}^{-1}$. (c) Single effect of DNP determined by two-dimensional screening at a AuNP concentration of 0 pM. (d) Single effect of DNP determined by one-dimensional screening.

measuring the increase of cell density by light scattering and the metabolic activity by microfluorimetric measurements of the endogenous cellular autofluorescence.

The effect-screening of the gold nanoparticle suspension (30 nm diameter) in microdroplets showed non-toxic effects on the autofluorescence signals of the *E. coli* bacteria for particle concentrations up to 10 pM (Fig. 3a). In ref. 10 the effects of functionalized 2 nm gold nanoparticles on an *E. coli* species had been specified with LC(50) values of 3 $\mu\text{mol L}^{-1}$ for cationic and >28 $\mu\text{mol L}^{-1}$ for anionic side chains of the functionalizing groups. In ref. 25 non-toxic effects were found for unmodified 10 nm gold nanoparticles up to concentrations of 1 mg L^{-1} .

The EC(50) value, defined as the concentration which causes 50% reduction in autofluorescence signal of *E. coli* compared with the controls, for the phenolic uncoupler DNP was determined to be $89 \pm 0.68 \mu\text{mol L}^{-1}$ by the two-dimensional screening (Fig. 3c) and $74 \pm 0.7 \mu\text{mol L}^{-1}$ by the one-dimensional screening (Fig. 3d) in microdroplets. Nearly the same effective concentration levels were thus found for both screening procedures.

In Fig. 4 the weak interactions of the binary mixture are shown for the growth, measured as optical density (Fig. 4a), and the autofluorescence (Fig. 4b) of the bacteria culture. Both diagrams show similar response of the cell culture as a function of the different concentration pairs of the effectors. The two-dimensional screening experiments allowed the simultaneous

† Normalized intensity = intensity signal/base signal.

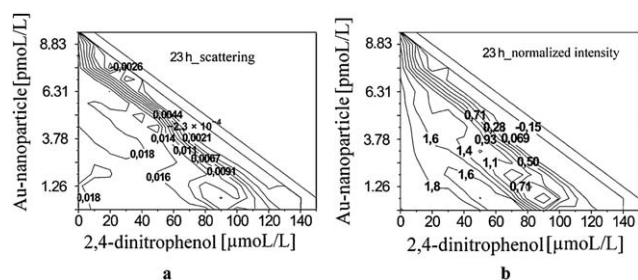


Fig. 4 Isoholes showing weak interactions up to zero interactions between the phenolic uncoupler DNP and gold nanoparticles. (a) Effects of the binary mixture on the growth (optical density) of *E. coli* after 24 h cultivation inside microdroplets. (b) Effects of the binary mixture on the autofluorescence of *E. coli* after 24 h cultivation inside microdroplets.

determination of a complete set of EC50 values for DNP within the whole range of investigated gold nanoparticle concentrations. The application of gold nanoparticles caused a reduction of critical concentration of DNP. Whereas the lethal concentration of DNP was about 100 pM in the absence of nanoparticles it is lowered to about 80 pM at a nanoparticle concentration of 3 pM and to 40 pM at a nanoparticle concentration of about 7 pM.

The results reflect a nearly linear (additive) behavior of DNP and the applied gold nanoparticle-containing suspension. Only a weak synergistic effect can be concluded from the experimental results, shown by the weak non-linear curves for low DNP concentrations.

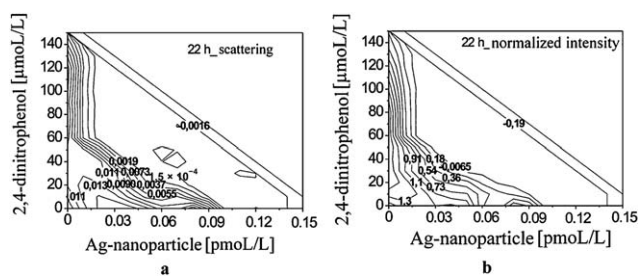
3. Microtoxicological screening of the binary mixture of silver nanoparticles and DNP inside microdroplets

The bacteriostatic effect of silver is well-known.²⁶ Despite the broad knowledge of the toxic effect of silver on cells and, in particular, on microorganism, to the best of our knowledge no detailed studies have been reported on the synergistic effects of silver nanoparticles and other toxins.

The critical value for impairing the growth of *E. coli* is much lower in the presence of silver nanoparticles as in the case of gold nanoparticles.

We found EC(50) values of about 90 ± 3 fmoL L⁻¹ for the investigated 20 nm silver nanoparticles (Fig. 3b). This value indicates stronger effects of the prepared silver nanoparticles on the viability of *E. coli* cultivated inside microdroplets as described by conventional methods.²⁷ The results are still in good agreement with the data reported in literature but strongly dependent on the investigated environmental matrix.^{28,29}

Concentrations between 0 and 150 fmoL L⁻¹ silver nanoparticles were investigated for their combinatorial effect with DNP in the concentration range between 0 and 150 μmol L⁻¹. For this purpose a micro-fluid segment sequence containing nearly 500 microdroplets was generated. They cover the whole two-dimensional concentration space of both effectors analogous to the experiment with DNP and gold nanoparticles. However, the observed results of the screening including the silver nanoparticles differ significantly from the investigations on gold nanoparticles. A strong synergistic effect was found for the binary mixture of AgNPs and DNP (Fig. 5).



mixture of DNP with the silver nanoparticle suspension a synergistic increase of the antibacterial effects was found. The strong interaction of silver nanoparticles in the presence of a phenolic uncoupler is presumably caused by released silver ions within the cell. The different pathways of endocytosis of metal nanoparticles might also play an important role for the mode of action in the bacteria cells.

Our results demonstrated the high potential of the microdroplet-based technique for the investigation of combinatorial effects in toxicology. Further developments of this powerful method may create an experimental approach to investigate highly resolved two-dimensional dose–response functions for a larger spectrum of toxins and drugs onto different cells and organisms. Eventually, this method will be relevant for future miniaturized screenings in environmental toxicology and drug development as well as related fields to gain detailed information on how cells respond to complex mixtures of substances.

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